

KULIKOV, K.A., doktor fiz.-matem. nauk, prof., nauchn. red.;  
SHUSTOVA, I.B., red.

[The universe around us] Vselennaia vokrug nas. Moskva,  
Znanie, 1965. 151 p. (Narodnyi universitet: Estestvenno-  
nauchnyi fakul'tet, no.12) (MIRA 18:12)

SHUSTOVA, I. F.

"Changes in the Cardiovascular System in Dysentery."  
Leningrad State Pediatric Medical Inst, Chair of Infectious Diseases  
in Adults and Chair of Therapeutics, Leningrad, 1955. (Dissertation  
for the Degree of Medical Sciences)

SO: M-955, 16 Feb 56

SHUSTOVA, I.F., assistant; VITKOVSKAYA, M.E., ordinator, BOBOMOLOVA, N.N.,  
vrach gorodskoy epidstantsii

Further observations on the treatment of dysentery in adults with  
furacilin and late results of an epidemiological investigation.  
Sbor. trud. Kursk. gos. med. inst. no.13:216-218 '58. (MIRA 14:3)

1. Iz kliniki infektsionnykh bolezney (zav. - dotsent M.Ye. Gal'perin)  
Kurskogo gosudarstvennogo meditsinskogo instituta.  
(DYSENTERY) (FURACILIN)

SHUSTOVA, K. (Astrakhan')

"Pedagogical lectures" in Astrakhan Province. Mat.v shkole no.1:  
82-83 Ja-P '56. (MIRA 9:4)  
(Astrakhan Province--Mathematics--Study and teaching)

SHUSTOVA, K.I. (Astrakhan')

Lessons for the analysis of tests. Mat.v shkole no.6:44-45 H-D '57  
(MIRA 10:11)

(Mathematics--Study and teaching)

SHUSTOVA, K.S.

Study of the effect of antibiotics of the tetracycline series on  
the capacity of E. coli to synthesize vitamin B<sub>12</sub>. Mikrobiol. zhur.  
26 no.5:57-60 '64. (MIRA 18:7)

1. Khar'kovskiy meditsinskiy institut.

L 54037-65 EWT(1) GS/GW  
ACCESSION NR: AT5010233

UR/0000/64/000/000/0112/0118

AUTHOR: Shustova, L. N.

TITLE: Determination of the external orientation elements Alpha and Omega using the Santoni solar periscope

SOURCE: USSR. Gosudarstvennyy geologicheskii komitet. Laboratoriya aerometodov. Spetsial'nyye voprosy fotogrammetrii (Special problems in photogrammetry). Moscow, Izd-vo Nauka, 1964, 112-118

TOPIC TAGS: strip set external orientation, solar goniometer, solar periscope, solar goniometer angle, photogrammetry, aerial surveying

ABSTRACT: In 1960, during a search for new power transmission line routes, the Laboratoriya aerometodov (Laboratory for Aerial Methods) carried out aerial surveying using the Santoni solar periscope. The camera of this periscope was rigidly connected with an AFA camera in such a way that the sun was photographed simultaneously with each aerial exposure. An appropriate processing of the pictures yielded the longitudinal  $\angle$  and transverse  $\omega$  inclination of the frame. The film processing was carried out with the Santoni solar goniometer, while the  $\angle$  and  $\omega$  angles were extracted by means of a solar calculator. The present paper outlines the method for the determination of angles  $\angle$  and  $\omega$  by means of the

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ACCESSION NR: AT5010233

solar periscope in considerable detail. Errors in mutual orientation obtained by differences of pairwise measurements were within  $\pm 3.0'$ . Orig. art. has: 12 formulas, 6 figures, and 4 tables.

ASSOCIATION: None

SUBMITTED: 17Oct64

ENCL: 00

SUB CODE: ES

NO REF SOV: 000

OTHER: 000

Card 2/2



Acceleration of the filtration process and its effect on the quality of water. P. V. Mozhukhin and L. N. Shustova. *Vodostokhenie i Snil. Tekh.* 1939, No. 3, 17-21; *Khim. Referat. Zhur.* 1939, No. 7, 92. An increase of the velocity of filtration from 5 to 8 g.5 m. hr. did not appreciably affect the physical chem. or the bacteriol. properties of the filtrate. The increase of the velocity of filtration lowered the total cost. W. R. H.

AS 56 SLA METALLURGICAL LITERATURE CLASSIFICATION

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The decolorizing of water in the Stalin water station.  
P. V. Muzhukhin and L. N. Shustova. *Vodosnabzhenie*  
*Soviet. Tekh.* 1939, No. 6, 34-41; *Khim. Referat. Zhur.*  
1939, No. 10, 94-5. The water of the Stalin water station  
is bacterially pure and is only slightly turbid, but humous  
substances give it a color and high O consumption. At-  
tempts were made to destroy the color of the water by frac-  
tional introduction of the coagulant, coagulation of a part  
of the water, introduction of the coagulant into the water  
in the form of a dry powder, addn. of the ppt. from the  
tank contg. a soln. of the coagulant, addn. of the ppt.  
coagulant, clay or lime, activated charcoal,  $KMnO_4$ ,  
 $FeSO_4$ , and  $Cl_2$  (with and without coagulant) and a pre-  
liminary chlorination. The best decolorizing agents are  
strong oxidizers, particularly  $Cl_2$ . The preliminary chlo-  
rination with 2.5-3 g./cu. m. of  $Cl_2$  is the most economical  
method and it permits decreasing the amt. of coagulant  
to 60 mg./l. Greater decolorization and greater saving in  
the coagulant can be effected by increasing the dose of  $Cl_2$   
to 6-8-10 g./cu. m., but with this quantity dechlorination  
is necessary. W. R. Hem

GAMOVA-KAYUKOVA, N. I.; SHUSTOVA, L. N.

Determination of stafilococci in food products. Gig. sanit.  
Moskva. no.9:33-36 Sept. 1950. (GIML 20:1)

1. Of the Central Sanitary-Hygienic Laboratory, Moscow.

PETROVICH, S.L.; SHUSTOVA, L.N.

Microflora of watermelons. Gig.sanit., Moskva no.3:41-44 Mar 1951.  
(CJML 20:7)

1. Of the Laboratory of the Sanitary Epidemiological Station,  
Moscow.

SHUSTOVA, L.N.  
SHUSTOVA, L.N.; LETROVICH, S.L.

Unification of the registration of coli bacilli in sanitation and  
bacteriological examinations of food products. Gig.i san. no.5:32-35  
My '54. (MIRA 7:5)

1. Iz Moskovskoy gorodskoy sanitarno-epidemiologicheskoy stantsii.  
(Escherichia coli)(Food—Bacteriology)

SHUSTOVA, L. N.

AID P - 3902

Subject : USSR/Medicine

Card 1/2 Pub. 37 - 6/21

Authors : Shustova, L. N. and S. L. Petrovich

Title : Methods of sanitary and bacteriological investigations of drinking water

Periodical : Gig. 1. san., 12, 23-26, D 1955

Abstract : Discusses the All-Union State Standard for specifications of quality of drinking water, issued May 1, 1954. This GOST 2874-54 is considerably changed when compared with the preceding GOST 2874-45, but presents the same methods of bacteriological water analysis as GOST 5216-50, four years its senior. The author recommends reviewing the standard methods of water analysis taking into consideration modern scientific literature and practical observations.

USSR/Microbiology, Sanitary Microbiology. Sanitary  
Microbiology of Food Products. 5-3

Abstr Jour : Ref Zhur - Biol., No 14, 1958, No 52359

Author : Shustova L.M., Petrovich S.L.  
Inst : Moscow Scientific Research Institute of Sanita-  
tion and Hygiene  
Title : On the Method of Sanitary Bacteriological Re-  
search in Milk and Milk Products

Orig Pub : Inform. byul. Mosk. n.-i in-t sanitarii i  
gigiyony, 1957, No 9, 39-42

Abstract : No abstract

Card : 1/1

IZRAIL'SKIY, V.P., prof.; doktor biolog.nauk; SHUSTOVA, L.N., kand.med.  
 nauk; GORLENKO, M.V., doktor biolog.nauk; MURAV'YEV, V.P.;  
 BEREZOVA, Ye.F., doktor biolog.nauk; SUDAKOVA, L.V., mikrobiolog;  
 GRUSHEVOY, S.Ye., doktor sel'skokhoz.nauk; NEMLIYENKO, F.Ye.,  
 doktor biolog.nauk; BEL'TYUKOVA, K.I., doktor biolog.nauk; STARYGINA,  
 L.P., kand.biolog.nauk; PERSHINA, Z.G., kand.biolog.nauk; ART'YEM'YEVA,  
 Z.S., mikrobiolog; NOVIKOVA, N.S., kand.biolog.nauk; OSNITSKAYA, Ye.A.,  
 fitopatolog; YASHNOVA, N.V., fitopatolog-mikrobiolog; MIKZABEK'YAN,  
 R.O., kand.biolog.nauk; TETUYEVA, I.V., red.; PEVZNER, V.I., tekhn.red.

[Bacterial diseases of plants] Bakterial'nye bolezni rastenii. Izd.2.,  
 perer. i dop. Moskva, Gos.izd-vo selkhoz.lit-ry, 1960. 467 p.  
 (MIRA 13:7)

1. Chlen-korrespondent Ukrainskoy AN (for Murav'yev).  
 (Bacteria, Phytopathogenic) (Plant diseases)



L 54035-65

EW(1)/FS(v)-3/ENG(v)/FSS-2 Po-4/Pe-5/Pq-4/Pae-2/P1-4  
TT/GS/GW

ACCESSION NR: AT5010231

UR/0000/64/000/000/0031/0036

AUTHOR: Shustova, L. N.

TITLE: The trajectory of the trace of the intersection between the optical axis and the surface of a celestial body

SOURCE: USSR. Gosudarstvennyy geologicheskii komitet. Laboratoriya aeromatodov. Spetsial'nyye voprosy fotogrammetrii (Special problems in photogrammetry). Moscow, Izd-vo Nauka, 1964, 31-36

TOPIC TAGS: optical axis, satellite photography, satellite camera inclination, weather satellite, satellite optics, satellite orientation

ABSTRACT: Depending on the orientation of the optical axis of a narrow-angle camera, a satellite may photograph different regions of a celestial body from otherwise identical orbits. Such a circumstance may partially explain the change in optical axis during proposed picture-taking from the Nimbus weather satellites as compared with the Tiros satellites (K. Ya. Kondrat'yev, Meteorologicheskiye sputniki, Gidrometeoizdat, M., 1963). Consequently, the determination of the trajectories of the trace of the intersections between optical axes and the surface of celestial bodies is of considerable interest. The author solves the problem by assuming that the orientation of the optical axis coincides ideally with a given

Card 1/2

L 54035-65

ACCESSION NR: AT5010231

direction. He considers three possible cases in which the optical axis is aligned 1) along a line between the sun and the celestial object, 2) perpendicular to the surface of the celestial object, and 3) towards the brightness center of the visible disk. Orig. art. has: 13 formulas and 3 figures.

ASSOCIATION: None

SUBMITTED: 17Oct64

ENCL: 00

SUB CODE: OP,SV

NO REF SOV: 004

OTHER: 000

Card 2/2

SHUSTOVA, L.Ye.

Density of rocks in the northeastern part of the Baltic Crystalline  
Shield. Geofiz.razv. no.13:72-81 '63. (MIRA 17:4)

TSIRUL'NIKOVA, I.Ya.; SHUSTOVA, L.Ye.; POROTOVA, G.A.

Deep-seated formations in the Pechenga structural zone  
according to geophysical data. Zap. LGI 46 no.2:14-16  
'63. (MIRA 17:6)

SHUSTOVA, L.Ye.

The Bothnia-Kandalaksha zone of a deep-laying trough in the earth's crust in the central part of the Baltic Shield. Dokl. AN SSSR 148 no.2:418-419 Ja '63. (MIRA 16:2)

1. Leningradskiy nauchnyy institut im. G.V. Plekhanova. Predstavleno akademikom A.A. Polkanovym.

(Bothnia Gulf region—Geology, Structural)  
(Kandalaksha Bay region—Geology, Structural)

L 05341-67 EJT(1)/FCC W  
ACC NR: AP7000236 SOURCE CODE: UR/0215/66/000/005/0047/0057

AUTHOR: Shustova, L. Ye.

ORG: Western Geophysical Trust (Zapadnyy geofizicheskyy trust)

TITLE: Deep structure of the Baltic shield determined from geophysical investigations

SOURCE: Sovetskaya geologiya, no. 5, 1966, 47-57

TOPIC TAGS: seismic prospecting, gravimetric survey

ABSTRACT: This is a summary of geophysical investigations carried out on the Baltic shield, both within the USSR and abroad. Accordingly, the author has made use of data from seismic, gravimetric and aeromagnetic investigations, aerial electrical prospecting, and other types of surveys for presenting a composite picture of current information on the shield. Figure 1 is a map of rock densities; Figure 2 is a gravimetric map; Figure 3 shows a map of crustal thickness; Figure 4 is a map of its block structure; Figure 5 is a map showing the rate of recent uplift of the shield; Figure 7 shows the block structure of the "granite" layer. The velocities of elastic waves in this area are constant to a depth of 5-8 km and therefore it is assumed that rock density remains constant to this depth. Beginning at 5-8 km the velocities increase to 6.5-7.0 km/sec, a value constant to a depth of 35-40 km. The density in the interval from 5-8 to 35-40 km is reckoned at 2.85-2.95 g/cm<sup>3</sup>.

Card 1/2

UDC: 551.1.(48)  
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L 05341-67  
ACC NR: AP7000236

The density of subcrustal matter is considered constant --  $3.3 \text{ g/cm}^3$ . The crust within the Baltic shield consists of two principal layers: an upper, or "granite" layer (to a depth of 5-8 km), characterized by a nonuniform composition, and a lower, or "basalt" layer, with a uniform composition and more basic in character. Orig. art. has: 7 figures. [JPRS: 37,658]

SUB CODE: 08 / SUM DATE: none / ORIG REF: 020 / OTH REF: 007

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Card 2/2

1ST AND 2ND ORDER		PRECISE AND PROPERTIES INDEX		1ST AND 2ND ORDER													
CA		<p><b>Determination of some maximum values (of the solutions).</b> N. A. Izmailov and M. B. Shustova. <i>Ukrain. Gosudarst. Inst. Eksp. Farm. (Kharkov: Kharkovskiy Materialy 1948, No. 2, 62-6.</i>—The surface tensions in dynes per cm. of quinine, cocaine, dionine, atropine and physostigmine in aq. soln. and their corresponding concns. in g. per 100 ml. of 0.1 N NaOH soln. were, resp.: 61.48 and 0.0005; 64.65 and 0.2722; 56.02 and 0.3500; 62.90 and 0.1528; 64.74 and 0.8442. The const. <math>\beta</math> and <math>1/\alpha</math> for these alkaloids in the Freundlich equation <math>\Delta\sigma = \beta \cdot c^{1/\alpha}</math> (<math>\Delta\sigma</math> is the difference between the surface tension of 0.1 N NaOH, and that of the alkaloid soln., <math>c</math> is the concn. of the soln. in mg. of alkaloid per 100 ml. of the solvent, <math>\beta</math> and <math>1/\alpha</math> are consts. detd. from the logarithmic graph) are, resp.: quinine 3.78 and 0.60, cocaine 1.43 and 1.1; dionine 1.61 and 1.61; atropine 2.48 and 0.49; physostigmine 0.830 and 1.17. The concn. of alkaloids can be detd. either directly from the surface tension-concn. graph or from the Freundlich equation given in the form <math>\lg c = (\lg \Delta\sigma - \lg \beta)/\alpha</math>. The concns. in mg. of alkaloid per 100 ml. of the solvent in the regions of greatest slope and the accuracy of the analysis are, resp.: quinine 10-20 mg. and 1.0%; cocaine 20-100 mg. and 8.0%; dionine 20-100 mg. and 4.0%; atropine 60-100 mg. and 1.7%; physostigmine 200-300 mg. and 5.0%. For detn. of alkaloids in mixts. with other medicinal substances the conditions of the detns. must be selected so that the remaining components have either a small effect on the surface tension, or their effect remains const.</p> <p style="text-align: right;">W. R. Henn</p>		17													
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION																	
<table border="1"> <thead> <tr> <th colspan="2">1ST ORDER</th> <th colspan="2">2ND ORDER</th> <th colspan="2">3RD ORDER</th> </tr> </thead> <tbody> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td> </tr> </tbody> </table>						1ST ORDER		2ND ORDER		3RD ORDER		1	2	3	4	5	6
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Shustova, M. B.

1520. Colorimetric determination of sulphide ions  
by the molybdenum thiocyanate reaction. V. A.  
Nazarenko and M. B. Shustova. *Zhur. Anal. Khim.*  
1956, 11 (4), 466-467. The instability of the  
complex between molybdenum thiocyanate and  
sulphides, which makes it unsuitable for color-  
metric determination of sulphides (Pepkowitz and  
Shirley, *Anal. Chem.*, 1951, 23, 1709), can be  
prevented by the addition of ethanol.  
G. S. Smith

Chem

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pm

*Shastova, M. B.*

7 5  
Colorimetric determination of sulfide ions by the molybdothiocyanate reaction. V. A. Nazarenko and M. B. Shastova. *Zhur. Anal. Khim.* 11, 489-91(1950). A mixt. of HCl solns. of  $(NH_4)_2MoO_4$  and KSCN can be used as a reagent for microquantities of  $S^{--}$  provided an org. solvent miscible with  $H_2O$  is added. The purpose of the org. liquid is to inhibit disocn. of the complex formed and to stabilize the color. M. Hoshikawa

*Am fra  
ang*

Shustova, M. B.

Distr: 4B4j

Colorimetric determination of sulfide ions by the molyb-  
dothioric reaction. V. A. Nazarcova and M. B.  
Shustova. J. Anal. Chem. U.S.S.R. 11, 617-19 (1950)  
(English translation).—See C.A. 51, 14460k. B. M. R.

*Shustova, M.B.*

AUTHORS: Nazarenko, V.A., Shustova, M.B.

32-11-3/60

TITLE: Analysis of Pure Metals. Determination of the Tantalum Content in Zirconium and Niobium (Analiz chistykh metallov. Opredeleniye primesi tantala v tsirkonii i niobii)

PERIODICAL: Zavodskaya Laboratoriya, 1957, Vol. 23, Nr 11, pp. 1283-1286 (USSR)

ABSTRACT: For the purpose of determining the tantalum content by the calorimetric method the derivatives of 2, 3, 7-trioxide-6-fluorine are recommended and dimethyl fluoron is particularly recommended. Tantalum in connection with dimethyl fluoron results in a bright red coloring. Without a content of tantalum the solution is yellow. Determination of the tantalum content at its minimum 3% is carried out from the 10 ml of the solution, which is decinormal with respect to hydrochloric acid, and 0.4% ammonium oxalate and contains 1 ml of the 1% gelatin solution. The following solutions are recommended for the processes of determination: 1. Mixtures of acids: a) 4-m nitric acid + 1-m hydrofluoric acid (70 ml of the 11-nitric acid + 118 ml water + 12 ml 40% hydrofluoric acid); b) 4-m with respect to hydrochloric acid + 2-m according to fluoric acid (70 ml of the 11-n nitric acid + 106 ml of water + 24 ml of 40% fluoric acid). 2. Dimethylfluoron:

Card 1/2

32-11-y/60

Analysis of Pure Metals. Determination of the Tantalum Content in Zirconium and Niobium

- 0.05% solution (50 mg + 0.5 ml of the 6-n nitric acid solution + 50 ml of 96% spirit). 3. Dilution solution: 10 g potassium pyrosulphate melt + 100 ml of the 4% solution of the ammonium oxalate + 250 ml water neutralized to slightly yellow by means of caustic potash. To this 50 ml of 2-n hydrochloric acid is added, and the entire mixture is dissolved in water up to 1000 ml. 4. Rinsing solution for extraction: 30 ml acid mixture as 1b + 20 ml ammonium sulphate solution + 20 ml isobutanol + 20 ml acetone. 5. Tantalum standard solution: 25 mg tantalum is dissolved in the mixture of fluoro- and nitric acid, after which 1 ml of sulphoric acid is added, and the whole is vaporized and then melted together with 2.5 g potassium pyrosulphate. The melt is dissolved in ammonium oxalate up to 250 ml. The paper then describes the process of determining the tantalum content in zirconium and in niobium. There are 4 tables.

AVAILABLE: Library of Congress

Card 2/2

SHUSTOVA, V.P., Cand Chem Sci-- (USSR) "Study of  
the diethylaminyl-1-fluorone as a new reagent for  
the <sup>detection</sup> ~~detection~~ of tantalum." Odessa, 1958, 11h pp  
in Higher Education USSR. Odessa State Univ in U.S.S.R.  
Technikov) 100 copies (81, 12-58, 113)

- 12 -

5(2), 5(4)

AUTHORS:

Nazarenko, V. A., Shustova, M. B.

SOV/32-24-11-9/37

TITLE:

Fluorometric Determination of Sulfate Ions and Spectrophotometric Determination of Thorium Using Derivatives of Trioxyfluoron (Fluorometricheskoye opredeleniye sul'fat-ionov i spektrofotometricheskoye opredeleniye toriya s pomoshch'yu proizvodnykh trioksifluorona)

PERIODICAL:

Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 11, pp 1344-1346 (USSR)

ABSTRACT:

Compounds such as the 9-(o-oxy-phenyl), 9-trichloro-methyl, and 9-propyl-2,3,7-trioxyfluorons can be used as complex-forming reagents for barium, thorium, and zirconium. Solutions of the unreacted fluorons, however, exhibit a tendency to fluoresce. Among the various trioxyfluoron derivatives available for the determinations mentioned in the title the 9-(o-oxy-phenyl)-trioxyfluoron (Salicylfluoron) appears to be the most suitable. This compound forms a red complex with thorium in weakly acidic medium ( $\text{pH} > 2$ ). The maximum light absorption of this complex lies at 500-530 m $\mu$  ( $\text{pH}=4.4$ ). The ratio of thorium to fluoron in the complex is 1:2. The measurements were carried

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SOV/32-24-11-9/37

Fluorometric Determination of Sulfate Ions and Spectrophotometric Determination of Thorium Using Derivatives of Trioxyfluoron

out on a Pulfrich (Pul'frikh) photometer after 24 hours on mixtures containing  $0.4 \cdot 10^{-5}$  to  $3.6 \cdot 10^{-5}$  moles Th and  $3.6 \cdot 10^{-5}$  to  $0.4 \cdot 10^{-5}$  moles salicylfluoron in 20% ethanol. The molar extinction coefficient of the salicylfluoron complex with thorium was found to be 26,000 at pH=4.4, 530 m $\mu$  and using  $0.3-1.0 \cdot 10^{-5}$  moles Th. The reaction obeys Beer's (Ber) Law. The determination of sulfate ion with salicylfluoronate is based on the formation of a sulfate complex which forms with the thorium complex, and according to the fluorescence of the free unreacted salicylfluoron the concentration of  $SO_4$ -ion can be determined. For the determination of microgram quantities of sulfate ion solutions of  $2 \cdot 10^{-4}$  molar thorium nitrate and  $5 \cdot 10^{-5}$  molar salicylfluoron are prepared. For quantitative determinations it is necessary to prepare a series of standard solutions, for example, with 0-0.25-0.5-1.0-1.5-2.0  $SO_4^{2-}$  sulfate ion. There are 2 figures and 5 references, 2 of which

Card 2/3



SOV/32-24-11-9/37

Fluorometric Determination of Sulfate Ions and Spectrophotometric Determination of Thorium Using Derivatives of Trioxyfluoron

are Soviet.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii Akademii nauk USSR  
(Institute of General and Inorganic Chemistry, AS UkrSSR)

Card 3/3

01.04

S/02/60/026/012/002/036  
B020/B056

55230

1273, 1350 only

AUTHORS:

Shustova, M. B., Nazarenko, V. A.

TITLE:

Analysis of Pure Metals. Determination of Vanadium  
Impurities in Titanium

PERIODICAL:

Zavodskaya laboratoriya, 1960, Vol. 26, No. 12, pp. 1339-1341

TEXT: In the present paper, the use of a method of determining vanadium quantities of less than one microgram (Ref. 1), which is based upon the catalytic acceleration of the aniline oxidation by potassium chlorate in the presence of oxine as activator (Ref. 2), is demonstrated by determining microquantities of vanadium in titanium. Under the conditions mentioned, the solution becomes yellowish-brown in the presence of vanadium, while otherwise the solution is light-yellow. The sensitivity of the reaction is increased by heating, but after a longer period of heating, dim solutions are formed, which cannot be photometrized. The reaction product may be extracted by means of organic solvents (ethyl or amyl acetates, isoamyl alcohol), in which case the extracts are brownish-red. During extraction of the reaction products, the detection limit is 0.01  $\mu$ -vanadium in 100 ml

Card 1/3

Analysis of Pure Metals. Determination of  
Vanadium Impurities in Titanium

87704

S. 032/60/026, 012/002, 036  
3020/3036

solution (maximum dilution  $1 : 10^{10}$ ). The light absorption curves of the ethyl acetate extracts obtained in the manner described in the absence and presence of 0.2% V are given in Fig. 1. They were recorded at the optimum wave length of 390 mμ. Fig. 2 shows the dependence of the optical density of the extracts on the quantity of vanadium during measurement in relation to the ethyl acetate by means of the spectrophotometer SF-4 (SF-4) at 390 mμ and by means of the horizontal photometer FM-56 (FMS-56) with the light filter MC-47 (MS-47) at 465 mμ. Larger quantities of titanium disturb, because they bind oxine; in quantities of up to 500 g, titanium may be masked by the addition of ammonium tartrate. In this case the sensitivity is reduced to one fifth. Up to 500 g iron may be masked by the addition of pyrophosphate without disturbing; also platinum does not disturb. The best results were obtained in the extraction with isoamyl alcohol. In this case vanadium can be quantitatively extracted at pH=5. Here, ammonium tartrate must, however, be added, which binds titanium to a complex; otherwise, the latter is precipitated. The results obtained show that by this method up to  $5 \cdot 10^{-5}\%$  V in 0.1 g titanium may be determined (Table). The method is not suited for analysis of titanium, which contains some tenths or hundredths of molybdenum. Molybdenum in quantities lower than

Card 2/3

37704

Analysis of Pure Metals. Determination of  
Vanadium Impurities in Titanium

3/032/60/026/012/002/036  
B070/B056

0.001% does not disturb the determination of vanadium. There are 2 figures,  
1 table, and 7 references: 4 Soviet, 1 Austrian, and 2 Japanese.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii Akademii nauk USSR  
(Institute of General and Inorganic Chemistry of the Academy  
of Sciences of the UkrSSR)

Card 3/3

24

S/137/62/000/053/112  
A160/A101

AUTHORS:

Nazarenko, V. A.; Shustova, M. B.

TITLE:

Determination of tantalum in lean ores by colorimetric means

PERIODICAL:

Referativnyy zhurnal, Metallurgiya, no. 3, 1962, 3, abstract 3 K 9.  
("Khim. fiz.-khim. i spektr. metody issled. rud redk. i rasseyan.  
elementov", Moscow, Gosgeoltekhizdat, 1961, 83 - 91)

TEXT:

It has been established that all trioxyfluoron derivatives can be used as reagents for Ta, yet the most sensitive and specific one of them is 9-paradimethyl aminophenyl-2, 3, 7-trioxy-6-fluoron, called dimethyl fluoron (I). The initial Ta water-base solution is evaporated to dryness in a Pt-bowl. The radical is subjected to a slight calcination, treated with 2 ml HF, evaporated to dryness, supplemented with 20 ml of a 5 %  $H_3BO_3$ , then again evaporated to dryness. Then it is melted at 500 - 600°C, supplemented with 5 g K persulfate and fused with it at 600 - 650°C. The melt is dissolved in 2.5 ml of a 4 %  $H_2C_2O_4$ , transferred into a 50 ml flask, neutralized for 4-dinitrophenol with the aid of 1 normal KOH solution until the appearance of a slightly noticeable yellow

Card 1/3

est tube is  
est (10 ml of the diluent,  
termination process is carried-  
was separated from inhibiting and  
earth "acids" with tannin, from a 3 - 6 %

Card 2/3

S/032/61/027/001/002/037  
B017/B054

AUTHORS: Nazarenko, V. A. and Shustova, M. B.

TITLE: Determination of Iodine Microimpurities in Elementary Silicon

PERIODICAL: Zavodskaya laboratoriya, 1961, Vol. 27, No. 1, pp. 15-16

TEXT: A method was developed to determine iodine microimpurities in silicon. The impurities are extracted with benzene after oxidation of the iodide to elementary iodine. The course of analysis is indicated: 1 or 0.5 g of finely ground silicon is dissolved in a 20-ml 3 N sodium hydroxide solution which is heated simultaneously. 5 ml of sulfuric acid 1:1 is added to the solution, and water is added until an amount of 150 ml is reached. The sample is placed in a separating funnel, mixed with sodium nitrite, and twice extracted with benzene. The iodine content is determined colorimetrically. Results are given in a table. By this method it is possible to determine 0.5  $\mu$  of iodine in 1 g of silicon, i.e.,  $5 \cdot 10^{-5}\%$ . This method is mainly intended for semiconductor silicon which contains small iodine impurities after production by the iodide method. There is 1

Card 1/2

Determination of Iodine Microimpurities  
in Elementary Silicon

S/032/61/027/001/002/037  
B017/B054

table.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii Akademii nauk  
USSR (Institute of General and Inorganic Chemistry, Academy  
of Sciences UkrSSR)

Card 2/2

NAZARENKO, V.A.; SMUSTOVA, M.B.; RAVITSKAYA, R.V.; NIKONOVA, M.P.

Determination of calcium, aluminum, and chromium impurities in  
antimony. Zav.lab. 28 no.5:537-539 '62. (MIRA 15:6)

1. Institut obshchey i neorganicheskoy khimii AN USSR.  
(Antimony--Analysis) (Metals--Analysis)



S/032/62/028/006/002/025  
B110/3101

AUTHORS: Mazarenko, V. A., Shustova, M. B., Shitareva, G. G., Yagnyatinskaya, G. Ya., and Ravitskaya, R. V.

TITLE: Determination of impurities in titanium

PERIODICAL: Zavodskaya laboratoriya, v. 28, no. 6, 1962, 645 - 648

TEXT: The determination of the contents of Ta, Al, P, Si, Mg, Cr, Mn, Fe, and Ni in Ti with an accuracy of 0.0001% is described. (1) Tantalum is photometrically determined with dimethyl fluorone (50 mg in 100 ml 96%  $C_2H_5OH$  and 0.5 ml 6 N HCl) after extraction as a fluorine complex with an acetone-isobutanol mixture. (2) Manganese is determined colorimetrically ( $HNO_3$ ,  $H_3PO_4$ , and potassium periodate) as manganic acid after extraction in the form of diethyl dithiocarbamate. (3) Iron is determined colorimetrically as thiocyanate after extraction of the oxinate (5 ml 1% oxine solution in 1 N  $CH_3COOH$ ) using chloroform in the presence of  $H_2O_2$  at  $pH > 8$ . (4) Nickel is colorimetrically determined with dimethyl glyoxime after the

Card 1/2

S/078/62/007/012/010/022  
B144/B180

AUTHORS: Nazarenko, V. A., Lebedeva, N. V., Biryuk, Ye. A., Shustova, M. B.

TITLE: Complex compounds of multivalence metals with trioxyfluorones

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 12, 1962, 2731-2738

TEXT: The complex formation between  $\text{GeO}_2$ ,  $\text{ZrOCl}_2$  or  $\text{SbCl}_3$  and phenyl fluorone and between  $\text{Sc}_2(\text{SO}_4)_3$  and propyl fluorone was studied spectroscopically in acid media after stabilization with gelatine to ascertain whether the metal ion substitutes two H atoms in the diphenol or one H atom in the o-hydroxyquinone. A new scheme, based on the solubility product, is given for the evaluation of the spectrophotometric data; this was necessary because of the low solubility of the complexes. The complex formation with Zr was studied in 0.2 - 0.8 N HCl and showed that only a 1:2 complex forms (optimum 0.2 - 0.3 N HCl). This was confirmed by both the isomolar series and the molar ratios. The Zr complex is thus consistent with other  $\text{M}^{\text{IV}}$  trihydroxy fluorone complexes. A study of the change in optical density as a function of the pH showed that only one H

Card 1/2

HAZARDOS, V.A.; LEBEDEVA, N.V.; SHUSTOVA, M.B.; BIRYUK, Ye.A.

Trihydroxyflirones. Metod.poluch.khim.reak. i prepar. no. 7:  
21-24. 1963. (MIRA 17:4)

1. Institut obshchey i neorganicheskoy khimii AN UkrSSR, Odessa.

SHUSTOVA, M.B.; NAZARENKO, V.A.

Trihydroxyfluorones as reagents for the photometric determination of molybdenum. Zhur.anal.khim. 18 no.8:964-971 Ag '63. (MIRA 16:12)

1. Institute of General and Inorganic Chemistry, Academy of Sciences, Ukrainian S.S.R., Laboratories in Odessa.

L 15203-65 EWT(m)/EPF(n)-2/EPR/EWP(b) Ps-4/Pu-4 SSD/ASD(a)-5/AFWL/ESD(gs)/  
ASD(f)-2/ASD(m)-3/AS(mp)-2 JD/JG/MLK

ACCESSION NR: AT4048100

S/0000/64/000/000/0150/0153

AUTHOR: Nazarenko, V.A., Shustova, M.B. B+1

TITLE: Photometric determination of microquantities of molybdenum in high-melting metals 27

SOURCE: Spektral'ny\*ye i khimicheskiye metody\* analiza materialov (Spectral and chemical methods of materials analysis); sbornik metodik, Moscow, Izd-vo Metallurgiya, 1964, 150-153

TOPIC TAGS: molybdenum, diethyldithiocarbamate, radiometry, quantitative analysis, colorimetric analysis, refractory metal, nitrophenylfluorone 27

ABSTRACT: For the separation of microquantities of molybdenum from difficulty fusible metals such as Ti, Zr, V, Nb, Ta and W, extraction with a chloroform solution of diethyldithiocarbamic acid was found to be more specific than methods based on molybdenum extraction after the addition of sodium diethyldithiocarbamate to a 1 N hydrochloric acid solution. The percentage of molybdenum extracted, determined radiometrically using  $\text{Mo}^{99}$  and photometrically with orthonitrophenylfluorone, amounts to 98% with extraction from 6 N  $\text{H}_2\text{SO}_4$  using phases of equal volume. Citric acid was used to retain the niobic, tungstic and other acids in solution. Vanadium was first reduced

Card 1/2

L 15203-65

ACCESSION NR: AT4048100

with tartrate to the tetravalent form. For the determination of molybdenum, its reaction with orthonitrophenylfluorone was used, the synthesis of which is described. The extraction of molybdenum is also described. This method makes it possible to determine  $5 \times 10^{-2}$  -  $2 \times 10^{-5}\%$  Mo in Ti, Zr, Hf, Nb and Ta, and  $3 \times 10^{-5}\%$  Mo in tungsten. The analytical results for the above-mentioned metals are tabulated. Orig. art. has: 1 table.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii Akademii nauk UkrSSR Institute of General and Inorganic Chemistry, Academy of Sciences Ukr. SSR)

SUBMITTED: 12Feb64

ENCL: 00

SUB CODE: MM, IC

NO REF SOV: 003

OTHER: 002

Card 2/2

L 52279-65 EWT(m)/EPF(n)-2/EWG(m)/EPR/ENP(t)/ENP(b) Ps-l/Pu-l IJP(c)

JD/JG

ACCESSION NR: AT5012673

UR/2513/65/015/000/0111/0120 28

AUTHOR: Shustova, M.B. 27

TITLE: Extraction of molybdenum in the form of the diethyldithiocarbamate complex (B+)

SOURCE: AN SSSR. Komissiya po analiticheskoy khimii. Trudy, v. 15, 1965. Metody kontsentrirvaniya veshchestv v analiticheskoy khimii (Methods of concentrating substances in analytical chemistry), 111-120

TOPIC TAGS: molybdenum extraction, molybdenum determination, refractory metal analysis, colorimetric analysis, diethyldithiocarbamate complex, orthonitrophenyl-fluorone 21

ABSTRACT: The author studied the stability of chloroform solutions of DDC acid (diethyldithiocarbamic acid), the absorption spectra of chloroform solutions of DDC acid and its complexes with molybdenum, and the dependence of the optical density of the extracts on the concentration of sulfuric acid in the aqueous phase and on the molybdenum concentration. The measurements were made with an SF-4 spectrophotometer. It was found that the concentration of chloroform solutions of DDC acid remains constant for 5 days if they are kept at a temperature not above 2C. Maximum

Card 1/2

L 52279-65

ACCESSION NR: AT5012673

optical density is displayed by the DDC complex of molybdenum extracted from 1 N sulfuric acid. A decrease in the optical density of the chloroform solutions of DDC complexes is due to the reduction of molybdenum by DDC acid to the pentavalent state. The Mo-DDC complex was found to have the composition  $\text{MoO}_2(\text{C}_4\text{H}_{10}\text{NCS}_2)_2$ , this being in agreement with the findings of other authors. The formation of this complex was used by the author to separate molybdenum prior to its determination in refractory metals - Ta, Nb, Ti, Zr, Hf, V, and W; the chloroform extract of the DDC-Mo complex was evaporated, the organic substances were decomposed by heating, and molybdenum was determined with orthonitrophenylfluorone. Orig. art. has: 6 figures, 2 tables, and 1 formula.

ASSOCIATION: Komissiya po analiticheskoy khimii, AN SSSR (Commission on Analytical Chemistry, AN SSSR)

SUBMITTED: 00

ENCL: 00

SUB CODE: IC, *SL*

NO REF SOV: 007

OTHER: 005

*Gah*  
Card 2/2.



ACC NR: AP6010053

SOURCE CODE: UR/0032/66/032/003/0267/0269

AUTHOR: Nazarenko, V. A.; Biryuk, Ye. A.; Shustova, M. B.; Shitareva, G. G.; Vinkovetskaya, S. Ya.; Flyantikova, G. V. 56  
B

ORG: Institute of General and Inorganic Chemistry, AN UkrSSR (Institut obshchey i neorganicheskoy khimii AN UkrSSR)

TITLE: Determination of impurities in tantalum 11

SOURCE: Zavodskaya laboratoriya, v. 32, no. 3, 1966, 267-269

TOPIC TAGS: tantalum, impurity level, photometric analysis, iron, copper, tin, lead

ABSTRACT: The photometric determination of impurities in tantalum is described. It has a sensitivity of  $10^{-4}\%$  and requires all the precautionary measures used during the analysis of high-purity metals, including the running of blank experiments under conditions of sample analysis. The photometric determination is preceded by extraction of the analyzed element (Pb, Cu, Fe, Ni, or Sn) from the tantalum sample, by extraction during the determination of tantalum in Zr, Bi, and Zn in the form of a fluortantalate complex, and by determination of chromium after separation of the tantalum by hydolysis. Lead and cadmium are determined by dithizone after extraction of the lead and cadmium (in the form of diethyldithiocarbaminates) from acid medium with chloroform. The interfering effect of other elements is eliminated by washing the extract with alkaline

Card 1/2

UDC: 543.7

ACC NR: AP6010053

solution (pH 12) containing cyanide, tartrate, and diethyldithiocarbamate. The rhodanide method, with extraction of the dyed complex, is used for the determination of iron. Copper is determined by dithizone. The separation of iron and copper from tantalum is made by extraction of their diethyldithiocarbamate salts. Tin is determined photometrically with paranitrophenylfluorone after extraction of the tin from the sulfate medium with chloroform in the form of diethyldithiocarbamate. This is made similarly to the determination of tin in niobium (N. B. Lebedeva, V. A. Nazarenko, Trudy Komissii po anaticheskoy khimii, Izd. AN SSSR, XI, 287, 1960). It is convenient to determine some impurities after separating the tantalum from them. This can be done by the extraction of the fluorotantalum complex with ketones (e.g., cyclohexanone) from its solution in HF and H<sub>2</sub>NO<sub>3</sub> or H<sub>2</sub>SO<sub>4</sub>, while Zr, Ti, Bi, and Zn can be determined in the aqueous phase: Zr with phenylfluorone, Bi by the iodide-ketone method, and Zn with dithizone. Chromium is determined with diphenylcarbazide after separation of tantalum by hydrolysis.

SUB CODE: 11,07/ SUBM DATE: none/ ORIG REF: 008

Card 2/2 hs

RAYKHSHTAT, G.N.; SHAPIRO, A.A.; SHUSTOVA, N.G.

Outbreak of whooping cough in a kindergarten. Zhur. mikrobiol., epid.  
i immun. 41 no.9:142 S '64. (MIRA 18:4)

1. Sanitarno-epidemiologicheskaya stantsiya Sverdlovskogo rayona  
Moskvy.

L 15673-66 EWT(m)/EWP(w)/T/EWP(t)/EWP(b) IJP(c) JD

ACC NR: AP6000196

SOURCE CODE: UR/0056/65/049/005/1431/1434

AUTHOR: Motulevich, G. P.; Shubin, A. A.; Shustova, O. F.ORG: Physics Institute im. P. N. Lebedev, AN SSSR (Institut fiziki AN SSSR)TITLE: The effect of periodic structure on the optical properties of aluminum

SOURCE: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 49, no. 5, 1965, 1431-1434

TOPIC TAGS: aluminum, optic property, refractive index, ir phenomenon, skin effect, conduction electron, electron collision, metal crystal, metal crystallization, light polarization, electron interaction, periodic system

ABSTRACT: The authors measured the real and imaginary parts of the refractive index of crystalline and amorphous aluminum in the infrared region. In both cases, layers of 99.99% pure aluminum were evaporated in vacuum on a glass substrate. A crystalline or amorphous structure was obtained by varying the cooling rate. The measurements were made by a polarization technique, using four-color reflection of light from the investigated surface, as described by the authors earlier (Optika i spektroskopiya, v. 3, 361, 1957). The measurements have shown that the skin effect exhibits a slightly anomalous character in crystalline aluminum at room temperature, but in amorphous aluminum it is almost normal. The concentration of the conduction electrons and the effective collision frequency of the electrons, which determine the refractive index, are calculated, and it is shown that on going from crystalline to amorphous layers, the conduction electron concentration increases from approximately

Card 1/2

SHUSTOVA, S.T.

Machine for removing the board from fabrics after pressing.  
Obm. tekhn. opyt. [MLP] no.11:43 '56. (MIRA 11:11)  
(Textile finishing)

PREOBRAZHenskAYA, I.N., inzh.; SHUSTOVA, S.T.

Innovators at Kuntsevo Textile Factory. Izobr.i rats. no.7:17-18  
Jl '58. (MIRA 11:12)

(Kuntsevo--Textile industry)

L 20597-66 EWT(d)/EWT(m)/EWP(w)/EWP(o)/EWA(d)/EWP(v)/T/EWP(t)/EWP(k)/EWP(l)/ETC(m)-6  
 ACC NR: AP6009808 JD (N) SOURCE CODE: UR/0096/66/000/004/0010/0013

AUTHOR: Elepko, V. F.; Shustova, T. A. (Engineer)

ORG: All-Union Heat Engineering Institute (Vsesoyuznyy teplotekhnicheskiy institut)

TITLE: Reliability<sup>14</sup> of austenitic steels in power units operating with 650C and 315 atm steam

SOURCE: Teploenergetika, no. 4, 1966, 10-13

TOPIC TAGS: austenitic steel, heat resistant steel, tube steel, steel property

ABSTRACT: Heat-resistant<sup>16</sup> austenitic steels EP17<sup>18</sup> and EP184<sup>14</sup> (both used in pipelines of the Kashira power station operating with steam 650C and 315 atm) were tested for the effect of prolonged aging<sup>16</sup> (up to 15,000 hr) at 550, 650, and 700C. Both steels, especially EP17, were found to undergo significant structural changes which affected their mechanical properties. At exposures up to 5000 hr, the structural changes<sup>16</sup> are limited to the precipitation of Cr<sub>23</sub>C<sub>6</sub> carbide and Fe<sub>2</sub>W intermetallic compound, with the precipitation of the latter becoming especially intensive after 3000, 5000, and 10,000 hr at 700, 650, and 550C, respectively. The precipitation of both phases continued for the entire test period (15,000 hr). After 10,000 hr, small amounts of Sigma-phase were observed and the notch toughness<sup>16</sup> of both steels dropped from the original 23—27 mkg/cm<sup>2</sup> to 8—10 mkg/cm<sup>2</sup>, regardless of the aging temperature. Prolonged aging also lowered the rupture strength, especially that of EP17 steel. In

Card 1/2

UDC: 669.15-194:621.772.4.001.45

L 20597-66

ACC NR: AP6009808

the first 3000—5000 hr, both steels develop a susceptibility to intergranular fracture which then disappears completely (EP184) or decreases (EP17) with prolonged aging. It is concluded that in service under the above conditions both steels, and especially EP17, are less reliable than earlier tested E1695 steel. Orig. art. has: 4 figures and 4 tables. [DV]

SUB CODE: 11/ SUBM DATE: none/ ORIG REF: 002/ ATD PRESS: 4224

Card 212 BK





AYUKHANOV, A.Kh.; VOSTRILOVA, N.V.; SHUSTROV, V.A.

Evaporation of the components of an oxide cathode in the course  
of its treatment. Radiotekh. i elektron. 7 no.9:1598-1607 S  
'62. (MIRA 15:9)

(Cathodes)

ARSENT'YEV, A.I.; YUMATOV, B.P., redaktor; SHUSTOVA, V.I., redaktor;  
MIKHAYLOVA, V.V., tekhnicheskij redaktor

[Earthwork by means of tractor and scraper units] Razrabotka  
mestorozhdenii traktorno-skrepnymi agregatami. Moskva, Gos.  
nauchno-tekhn. izd-vo lit-ry po cherno i tsvetnoi metallurgii,  
1955. 135 p. (MLRA 8:6)  
(Earthwork)

SNISARENKO, L.I.; CHIVIR'OV, O.M. [Chyvyr'ov, O.M.]; POZNYAKOVA, L.Ye.  
[Pozniakova, L.IE.]; SHUSTOVA, V.P.

Sanitary and hygienic work conditions in the tin can shops  
of canned food enterprises. Khar. prom. no.4:34-36 O-D '65.  
(MIRA 18:12)



SHUSTOVA, Ye.A. ....

Studying the dormancy of tree seeds. Uch. zap. Sar. un. 64:139-  
143 '59. (Trees) (Germination) (MIRA 13:9)

SHUSTOVA, Ye. A.

Cand Biol Sci - (diss) "Explanation of the causes for slow growth of seeds (fruit) of several tree varieties." /Kazan'7, 1961. 18 pp; (Ministry of Higher and Secondary Specialist Education RSFSR, Kazan' Order of Labor Red Banner State Univ imeni V. I. Ul'yanov-Lenin); 180 copies; price: free; (KL, 10-61 sup, 211)

15

Determination of the amount of phosphoric acid required by soils. Yu. K. Kudrin and R. N. Shostova. *Russk. Zapiski Tselovoi Prom.* 10, No. 28, 63-8 (1933). — Of the six methods compared for the detn. of the amt. of  $P_2O_5$  required by different soils the most accurate was the one of Truog. During the run of a test the room temp. should not vary more than  $2-3^\circ$ . V. E. Balkow



C 1

13

Manganese content in the soil and in plants on prolonged application of fertilizers. Yu. K. Kudzin and E. M. Shustova. *Doklady Akad. Nauk USSR*, R.S.R. 1936, No. 2, 167-70; cf. C. I. 44, 10246. N ( $(\text{NH}_4)_2\text{SO}_4$ ) and P (superphosphates) proved to be most effective in mobilizing Mn in soil when applied to sugar-beet crops on chernozem.

K (KCl) showed little effect, alone or in combination. Each of the minerals increased Mn in leaves, the N-K combination giving highest increase especially in old leaves. In wheat, the P-N combination was most effective on leaves whereas P alone gave highest Mn content in the harvested grain. Murtya Senkus.

SHUSTOVA, Ye. N.

"Conditions of Phosphate and Calcium Nourishment of Sugar Beets  
During Crop Rotation." Sub 6 Dec 51, All-Union Sci Res Inst of  
Fertilizers, Agricultural Engineering and Soil Science.

Dissertations presented for science and engineering degrees in  
Moscow during 1951.

SO: Sum. No. 480, 9 May 55

SHUSTOVA, E.N.

MD The influence of a layer system of plowing on several physical and biochemical properties of the soil. E. N. Shustova and Yu. I. Usenko. *Pochvedenie* 1955, No. 12, 43-51. The system of layer type of plowing (0-15 cm. is turned over into the furrow, 15-30 cm. is simply shattered and remains in place, and the lower layer, 30-45 cm. is brought up to the surface) causes a rise in nitrate N during the first year as compared with the conventional form of plowing. After 3-4 years the 30-45 cm. layer brought to the surface behaves like the regular plowed layer in terms of nitrification. The original plowed-under 0-15 cm. layer does not lose its nitrifying power. The result is that with this system of plowing the nitrate N rises. J. S. Joffe

L 28529-66 EWP(j)/EWI(m)/EWP(t)/ETI IJP(c) RM/JD/WB/GD

ACC NR: AT6013803

(A)

SOURCE CODE: UR/0000/65/000/000/0284/0295

AUTHOR: Rozenfel'd, I. L.; Persiantseva, V. P.; Reyzin, B. L.; Shustova, Z. F.; Gavrish, N. M.

ORG: none

TITLE: Investigation of certain nitrobenzoic amine salts as corrosion inhibitors for ferrous and nonferrous metals

SOURCE: Korroziya metallov i splavov (Corrosion of metals and alloys), no. 2. Moscow, Izd-vo Metallurgiya, 1965, 284-295

TOPIC TAGS: amine salt, corrosion inhibitor, ferrous metal, nonferrous metal

ABSTRACT: The article presents the results of an investigation of the protective properties of certain inhibitors (nitro- and dinitrobenzoates) synthesized at the authors' laboratory; these properties were tested in natural as well as accelerated conditions involving cyclic and continuous exposure to moisture, with the aid of a specially developed device (Persiantseva, V. P., Rozenfel'd, I. L. Zavodskaya laboratoriya, 1958, 24, 7, 282). (The tests under natural conditions simulated the conditions under which metal products are stored in unheated warehouses and lasted for 21 months.) The inhibitors investigated were: hexamethyleneimine meta-nitrobenzoate, hexamethyleneimine ortho-nitrobenzoate, hexamethyleneimine 3,5-dinitrobenzoate, and piperidine 3,5-dinitrobenzoate. The coating of metal surface with

Card 1/2

L 28529-66

ACC NR: AT6013803

an inhibitor was accomplished through adsorption from vapor phase or by washing the specimens in alcohol solutions of the inhibitors with subsequent drying at room temperature. Protective properties were determined according to the time elapsed until the appearance of first signs of corrosion and according to corrosion rate (as determined by gravimetric method). Findings: When applied in the form of alcohol solutions, all the four tested chemicals proved to be effective inhibitors of atmospheric corrosion under conditions simulating storage of metals in unheated warehouses, in industrial districts (where the atmosphere is more contaminated), for not only ferrous metals but also the most widely used nonferrous metals, (Cu and its alloys, Ag, Sn, Al and its alloys, Ni and Cr coatings, and Zn and Cd coatings passivated in a  $K_2Cr_2O_7$  solution). These findings should represent a major advance considering that previously the only other known volatile inhibitors used in industry protected only ferrous metals. Orig. art. has: 7 tables and 1 figure.

SUB CODE: 11, 07 / SUBM DATE: 19Jul65/ ORIG REF: 004/ OTH REF: 002

Card

2/2

cc

SHUSTOVSKIY, F. A.

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USSR/Medicine (Veterinary) - Tissue Therapy Apr 52

"Experience in the Use of Preparation ASD," F. A.  
Shustovskiy

"Veterinariya" Vol XXIX, No 4, pp 49-51

The activity of ASD was tested by military-veterinary hospitals in various diseases on 138 animals. Two types of ASD were applied intravenously, per os (for intestinal diseases), or externally (for the treatment of wounds). ASD was found to be an extremely effective tissue therapy prep and stimulant.

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*SHUSTOVSKIY F.A.*

ALICHKIN, S.L.; AGRINSKIY, N.I.; ANDREYEV, G.F.; BAKUMENKO, G.D.;  
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IVANOVTSSEV, P.V.; KINBURG, M.Ya.; KOVALEV, P.A.; KOZLOVSKIY, Ye.V.  
KORNIYENKO, A.P.; KOLYAKOV, Ya.Ye.; LAKTIONOV, A.M.; LEVADNIY, B.A.  
MEDVEDEV, I.D.; NOVIKOV, N.V.; ORLOV, F.M.; OSTROVSKIY, A.A.;  
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(nachal'nika - prof. G.S. Pervomayskiy) Voenno-meditsinskiy  
ordena Lenina akademii imeni S.M. Kirova.

1. The first part of the document is a list of names.

2. The second part of the document is a list of names. The names are not clearly visible.

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Cinophotomicrographic study of *Toxoplasma gondii* in a peritoneal  
exudate of white mice. Dokl. AN SSSR 165 no.5:1215-1216 D '65.  
(MIRA 19:1)

1. Submitted February 8, 1965.

ACC NR: AP7008115

SOURCE CODE: UR/0020/67/172/004/0835/0888

AUTHOR: Zandberg, E. Ya.; Rasulev, U. Kh.; Shustrov, B. N.

ORG: Physicotechnical Institute im. A. F. Ioffe, Academy of Sciences, SSSR (Fiziko-  
tekhnicheskii institut Akademii nauk SSSR)

TITLE: Thermionic emission of positive ions of certain organic compounds from tung-  
sten oxides

SOURCE: AN SSSR. Doklady, v. 172, no. 4, 1967, 885-888

TOPIC TAGS: thermionic emission, tungsten compound

ABSTRACT: Experiments were carried out on thermionic emission from tungsten oxides in a mass spectrometric apparatus in the presence of various organic compounds at 10-5 mm Hg. The following compounds produced thermions: diethylamine, phenol, aniline, trimethylhydrazine, acetone peroxide, several amino acids, and also acetic and formic acid. Most attention was devoted to the ionization of the first four compounds. The spectra of thermionic emission from tungsten oxides (at  $T \leq 1100^\circ\text{K}$ ) and tungsten (at  $T \geq 2000^\circ\text{K}$ ) are tabulated. With the exception of aniline, ions representing products of surface reactions were observed in all cases. The results are in accord with previously advanced hypotheses on the formation of thermions by both catalytic dissociative ionization and formation of "heavy" ions in chemical surface reactions. The temperature dependence of thermionic currents from tungsten oxide

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surfaces was determined; the bell-jar shape of the  $I = f(T)$  curves obtained indicated the simultaneous occurrence of ionization and dissociation of the particles on the surface. In the case of aniline, the  $I = f(T)$  function was exponential. It is noted in conclusion that the thermal ionization of organic compounds on the surface of solids may be used as a method of studying processes of heterogeneous catalysis. Authors thank N. I. Ionov for discussing the results and I. N. Bakulin for his assistance. The paper was presented by Academician Konstantinov, B. P., 13 Apr 66. Orig. art. has: 3 figures and 1 table.

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57-6-29/36

AUTHOR: MAMYRIN, B.A., SHUSTROV, B.H.

TITLE: Mass-Spectrometer with Resolving Power of the Order of Several  
Thousands. (Mass-spektrometriya s razreshayushchey siloy v  
neskol'ko tysyach, Russian)

PERIODICAL: Zhurnal Tekhn. Fiz., 1957, Vol 27, Nr 6, pp 1347 - 1356  
(U.S.S.R.)

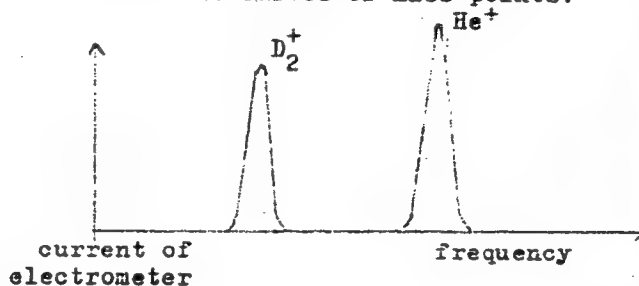
ABSTRACT: It is shown that a pulse-resonance-mass-spectrometer can be used  
as an analytical device for gas-analytical purposes with a re-  
solving power of several thousands. In order to realize this  
possibility the light intensity and sensitivity of the device  
were considerably increased 1) by the method of collecting ions  
in the source, 2) by the application of a specially developed  
generator of millimicroseconds-pulses with an increased sequence  
of frequencies, 3) by clarifying the basic causes of the oc-  
currence of a remaining current and elaboration of a measuring  
system for its removal. With respect to the production and ad-  
justment the device developed is more simple than those with  
double focussing and with a non-uniform field. One of the ad-  
vantages offered by the device is the possibility of regulating  
the resolving power without effecting any changes in the vacuum  
chamber. This is possible by selecting the suitable number of  
revolutions by the frequency transformation of generator pulses.

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Mass-Spectrometer with Resolving Power  
of the Order of Several Thousands.

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In the case of a reduction of the resolving power the light intensity of the device is increased. An important property of the device when used for purposes of analyses is the lack of "tails" or "trains" on the basis of the curves of mass points. (With 8 illustrations and 6 Slavic references)



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SUBMITTED: 3.3.1957

AVAILABLE: Library of Congress

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